

INQUIRY LAB Z

WHAT IS UP WITH ALL OF THESE WHITE POWDERS?

Note: It is not appropriate for you to look-up how to do this lab online or in books. You should be able to simply consult your class notes and your textbook. It is also inappropriate to consult the help of anyone who is not a member of your lab group.

Purpose:

What you are going to do? Determine the identity of all of the samples of white, powdery solids.

How are you going to do it? Well, that is the entire purpose of the lab, isn't it?

The Unknowns (these will be randomly labeled A-H):

Ammonium nitrate
Glucose

Calcium carbonate
Magnesium oxide

Sodium acetate
Sodium sulfate

Barium nitrate
Potassium iodide

Theory:

You will be given eight beakers of white solids. You will develop a flow chart type procedure that will systematically, step by step, test for certain characteristics. These characteristics will reveal the identity of the eight solids. This flowchart should identify the procedural test performed and what it reveals or fails to reveal.

Equipment/Materials Provided:

magnifying glass
beaker tongs
metal ring
centrifuge
stirring rods
filter paper
scupula
burets
conductivity tester
Benedict's Reagent

thermometer
beakers
flasks
test tubes
milligram balance
drying oven
spatula
separatory funnel
wire loop
0.100 M NaOH

crucibles
test tube clamps
evaporating dish
pipettes
mini-well plates
mortar/pestle
Petri dishes
plastic funnel
Bunsen burner
0.100 M HCl

laboratory tongs
ring stand
well plates
pipetter
Buchner funnel
graduated cylinders
distilled water
pH paper
splints
0.100 M AgNO₃

Preparation Week:

You will be given this document three weeks prior to the week of 2/26/18 through 3/02/18 (LAB A). Take this time to get together with your lab partners and create a plan to address the problem outlined above.

Week 1 (2/12/18 through 2/16/18):

Report to 5213 this week for lab. Here you will present your tentative procedure to your instructor. You must also include a data table for all critical measurements (qualitative or quantitative) and display how you will use this data, logically. This will be somewhat of a rough draft. **Please TYPE the procedure for legibility. Include a flowchart with your procedure.** Do not write your finalized procedure for Week 2 in your lab notebook until it is completely done and completely vetted. In lab this week you will receive suggestions to better meet your goals and to fine-tune your procedure.

Week 2 (2/26/18 through 3/02/18):

Come to lab with a finalized and formal purpose, illustrated procedure, flow chart and safety/physical data and a data table for your experiment. Perform your experiment. Find out the actual identity of the eight solids. Complete the lab (create an expanded conclusion that highlights the overall procedural steps leading to the identity of the unknowns, error analysis, post-lab questions, etc.).

Post Lab Questions:

1. If you were able to perform this lab again, what is the largest change you would make to your procedure? Justify your answer.

2. What equipment could you use to improve your experiment that was not available on the list? Justify your choice.
3. Do you think that you benefited from the opportunity of designing and performing this lab? Was it worth a 2 week commitment?